

## I Semester M.Sc. Examination, January 2017 (Semester Scheme) (NS) (2010-11 Scheme) CHEMISTRY

C-102 : Organic Chemistry – I

Time: 3 Hours Max. Marks: 80

Instruction: Answer question number 1 and any five of the remaining.

1. Answer any ten of the following:

 $(10 \times 2 = 20)$ 

- a) What is Huckel's rule? Explain the rule using tropane as an example.
- b) Draw the potential energy diagram for the following reaction.

c) Explain how isotope labeling is used to determine the mechanism of the following reaction.

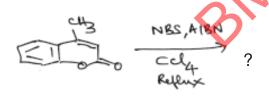
- d) What are ambident nucleophiles? Give any two examples.
- e) Draw the most stable conformations for the following with justification.



f) Indicate (if any) prochiral groups/faces in the following.

- g) What is Strecker reaction? Explain with an example.
- h) Suggest a suitable reagent and propose a mechanism for the following.

- i) What is Kiliani-Fischer synthesis? Explain with a suitable mechanism.
- j) Draw the conformational structures of D-and L-forms of mannose.
- k) Give the synthesis of purine from uric acid.
- I) Predict the product and propose a mechanism.



- 2. a) Write on aromaticity in naphthalene and any one meso-ionic compound.
  - b) Give an account of generation, structure, stability and reactivity of benzyl cation.
  - c) According to CIP rules, write the R/S configurational notations of the following.



3. a) Explain how the presence of intermediate is helpful in establishing the mechanism of the following reactions.

- b) What is Cram's rule? Explain with a suitable example.
- c) Write a note on conjugation and hyper conjugation in organic compounds.

 $(3 \times 4 = 12)$ 

4. a) Discuss the effect of resonance on the acidity and basicity of the following pairs.



b) Give an account of nomenclature and conformations of



c) Suggest a suitable synthetic scheme for the following dipeptide: (3×4=12)

- 5. a) List out the major differences between the  ${\rm S_N1}$  and  ${\rm S_N2}$  , mechanisms.
  - b) Give a brief account of the conformations and stability of 1,3-dimethylcyclohexanes.



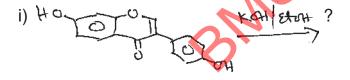
- c) Write a note on:
  - i) Racemization in peptide synthesis
  - ii) Peptidomimetics

 $(3 \times 4 = 12)$ 

- 6. a) Explain the method of determination of configuration of D-glucose.
  - b) Outline the solution phase synthesis of Leu-Enkephalin.
  - c) Predict the products

ii) 
$$\lim_{k \to \infty} \frac{cH_3T/koH}{2} ? \xrightarrow{A} ?$$
iii)  $\lim_{k \to \infty} \frac{Na_2S_2O_3}{2} ? \frac{ka_0 \cdot NH_3}{2} ? (3\times 4=12)$ 

7. a) What are the products of the following reactions?



- b) Outline the synthesis of aldonic acid.
- c) Discuss the Sangers method of sequencing the following peptide.
   GlyAlaPheGly. (3×4=12)